



Minnesota Minerals Education Workshop  
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# Mineral and Rock Identification

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# Introduction to Rock Forming Minerals

All minerals meet these six criteria:

1. **Naturally occurring** (not man-made)
2. **Inorganic** (no plant matter)
3. **Homogeneous** (must be the same throughout)
4. **Solid** (no liquids)
5. **Definite (but variable) chemical composition**
6. **Ordered internal arrangement of atoms**

# Identifying Minerals

- Many rock properties can be assessed very easily in a classroom
  - Color
  - Streak
  - Hardness
  - Specific Gravity

Note: Where applicable, image citations are listed in the presentation notes for each slide

# Mineral Properties: Color

- Color comes from the chemical composition of a mineral
- Easy to identify, but ***not always reliable***
  - Some minerals have an inherent color and are always the same
  - But some minerals come in a variety of colors based on presence of trace elements



Exhibit A:  
All  
fluorite!  
←











# Mineral Properties: Streak

- Streak is the color a mineral produces when ground to a fine powder
- ***Much more reliable than mineral color,*** because each mineral has a diagnostic streak color, no matter the color of the mineral **itself**
- Note: ***many minerals do not produce a streak***
- Test streak by marking the mineral on a white porcelain streak plate - ***press firmly!***



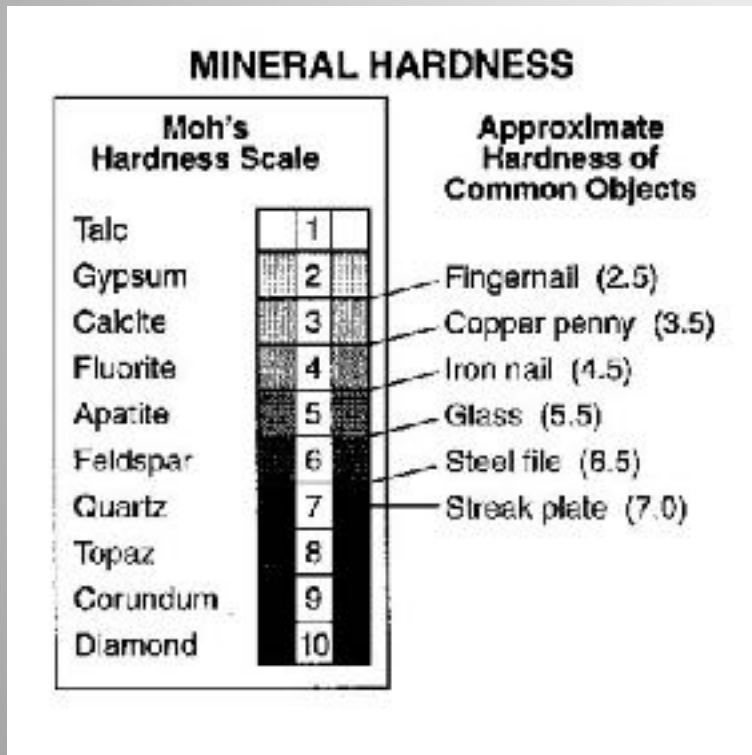
# Mineral Properties: Hardness

- The hardness of a mineral is a factor of the strength of the chemical bonds within the mineral
- Hardness is shown by a mineral's ability to resist or inflict abrasion (a scratch) on or by another mineral
- Hardness is measured from **1 (softest) through 10 (hardest)** on the Moh's Hardness Scale
- Hardness is tested by scratching a mineral against various reference materials

Mineral	Moh's Hardness	Image
Talc	1	
Gypsum	2	
Calcite	3	
Fluorite	4	
Apatite	5	
Orthoclase	6	
Quartz	7	
Topaz	8	
Corundum	9	
Diamond	10	

# Hardness Test

- Use common classroom materials to test the hardness of minerals



- If a mineral can be scratched by your fingernail, its hardness is:  
~~Less than~~  
2.5
- If a mineral scratches a streak plate, its hardness is:  
~~Greater than~~  
7.0
- If a mineral can scratch glass but can be scratched by a steel file, its hardness must be:  
~~Between 5.5 and~~  
6.5

# Mineral Properties: Specific Gravity

- Specific gravity of a mineral is its density ( $\rho$ ) relative to water
  - Specific gravity = 
$$\frac{\rho_{mineral}}{\rho_{water}}$$
- In a classroom, you can measure specific gravity by doing a density test or by a more subjective “heft test”
  - “Heft test”: Does the mineral feel heavier, lighter, or about what you’d expect based on its size?
    - Heavier = high specific gravity
    - Lighter = low specific gravity
    - About what you’d expect = medium specific gravity



# Mineral Properties: Magnetism

- Magnetism is the ability of some minerals to behave like magnets
- Magnetism is strongly related to iron content

Assess magnetism by holding a magnet up to a sample and feeling if there is a “pull” when you move the mineral farther away.

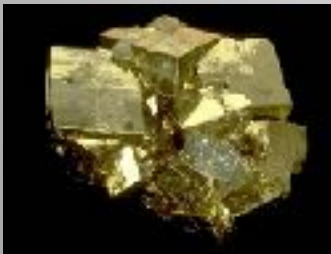


# Mineral Properties: Luster

- Luster refers to how a mineral reflects light
- Mineral luster is classified generally as “metallic” or “non-metallic”
  - *Metallic*: looks like a metal
  - *Non-metallic*: does **NOT** look like a metal
    - ***Caution: can still be shiny!***
    - Glassy, shiny, greasy, waxy, dull, earthy

# Luster (continued)

- Metallic



- Non-metallic





















# Mineral Properties: Cleavage

- Cleavage describes the way a mineral breaks (cleaves) along planes of weakness in its atomic structure
- ***Caution: Not all minerals have cleavage.*** Some minerals break or fracture in irregular ways
- Cleavage does form:
  - Flat, reflective surfaces
  - Angular corners
  - Defined edges
- Cleavage does NOT form:
  - Lumps & bumps
  - Irregular surfaces
  - Conchoidal fracture

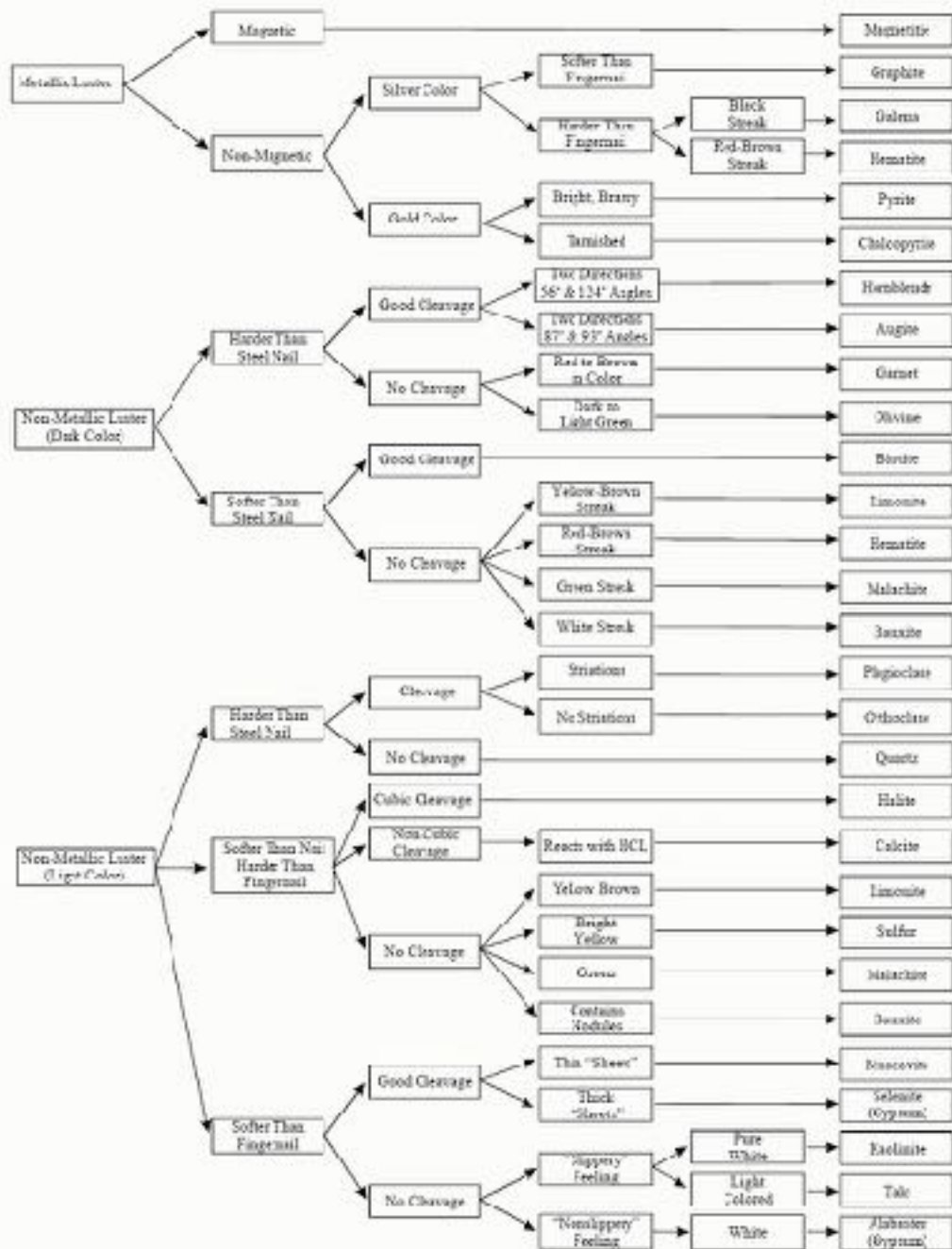


# Cleavage (continued)

- Minerals can have one, two, three, or even four different directions of cleavage depending on the strength or weakness of a mineral's atomic bonds.
- Multiple cleavage faces intersect each other at specific and predictable angles in certain minerals

Number of Cleavage Directions	Sketch	Illustration of cleavage directions	Example
1			
2 at 90°			
2 not at 90°			
3 at 90°			
3 not at 90°			
4			

### Mineral Identification Flowchart

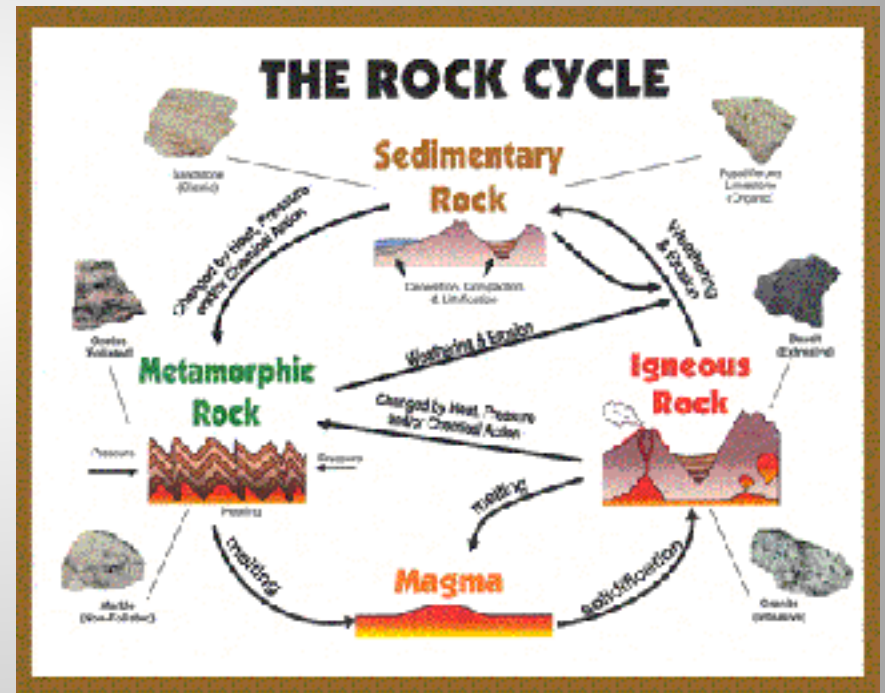


# So, how many minerals should students be familiar with?

- Important rock-forming minerals
  - Quartz
  - Plagioclase
  - Orthoclase
  - Hornblende
- Minerals with recognizable and diagnostic properties
  - Sulfur
  - Pyroxene
  - Pyrite
  - Olivine
  - Galena
  - Biotite
  - Magnetite
  - Muscovite
  - Hematite

# Minerals to Rocks

- *All rocks are made up of minerals*
- *Most rocks are composed of numerous minerals*
- Three types of rocks
  - Sedimentary
  - Igneous
  - Metamorphic
- These three rock types are interconnected through the rock cycle. Erosion & weathering, heat, and pressure can transform any rock types into another, given enough time





# Sedimentary Rocks

Sedimentary rocks form by deposition of sediment or precipitation of minerals at or near Earth's surface.

- Two types of sedimentary rocks
  - **Clastic**: form by cementing together fragments or grains of pre-existing rocks
  - **Chemical**: form when minerals precipitate out of water solutions at the earth's surface



# Common Sedimentary Rocks

## *Clastic*



Sandstone



Conglomerate/Breccia

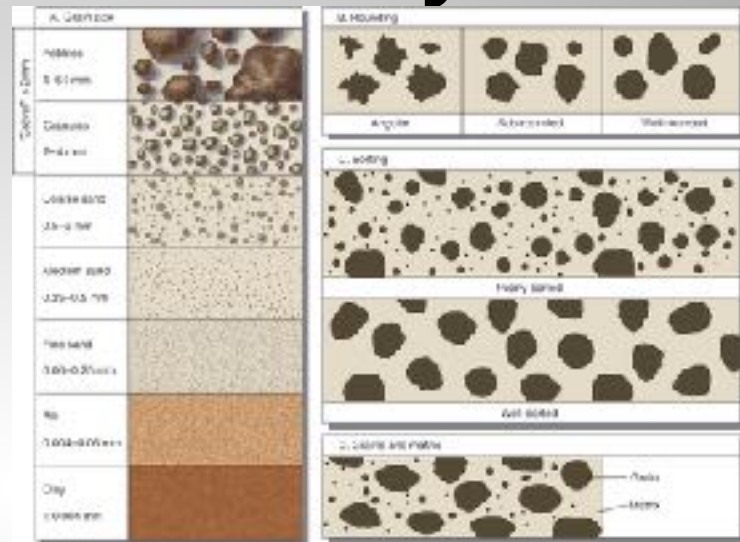
## *Chemical*



Limestone

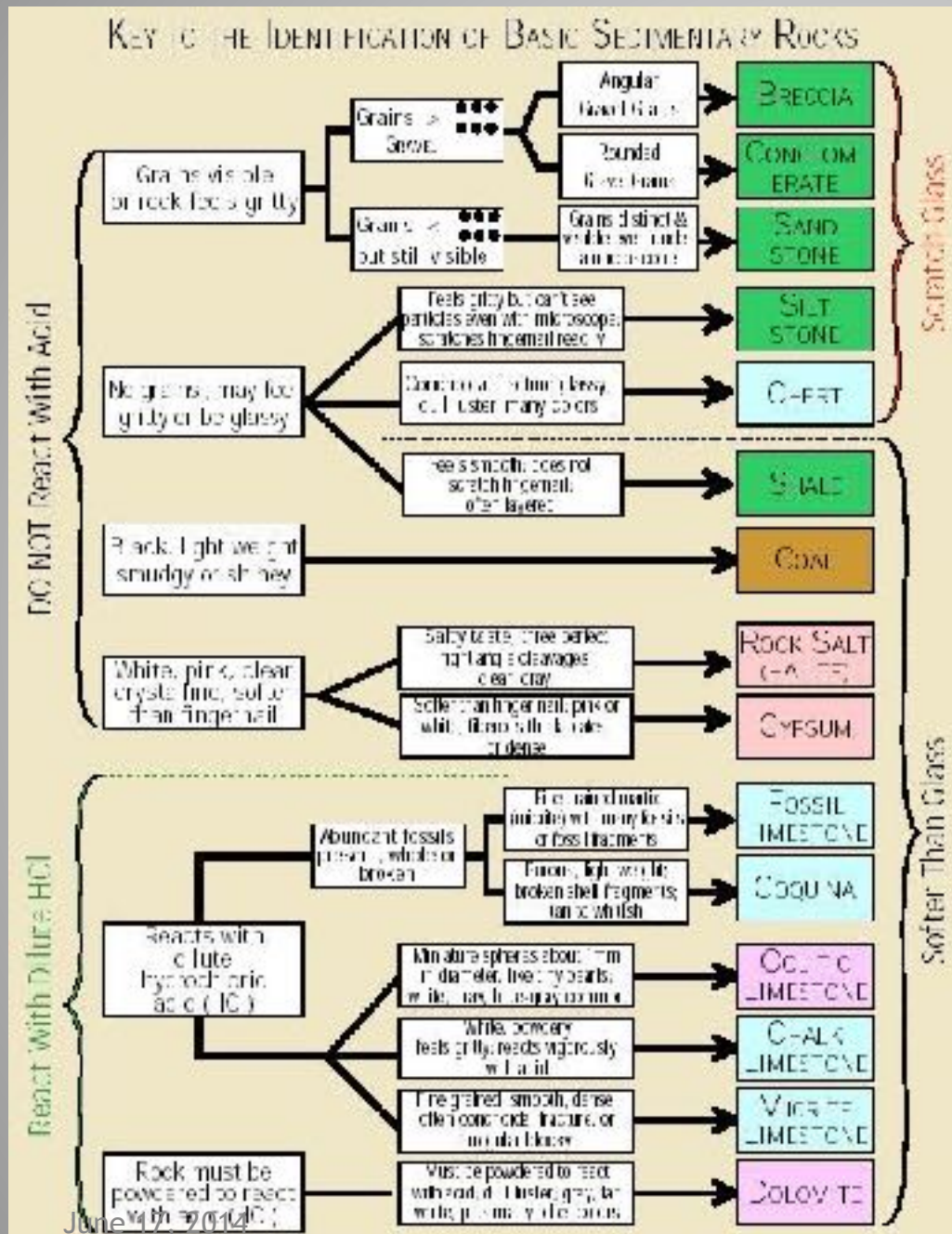
# Identifying Sedimentary Rocks

- Grain Size
  - Fine, medium, or coarse-grained
- Grain Sorting
  - Well or poorly sorted
- Grain shape
  - Rounded or angular
- Reaction with dilute HCl (Hydrochloric acid)
  - Fizzes readily, fizzes after being powdered, or does not fizz
- Other “dead-giveaway” characteristics
  - Fossils
  - Layering



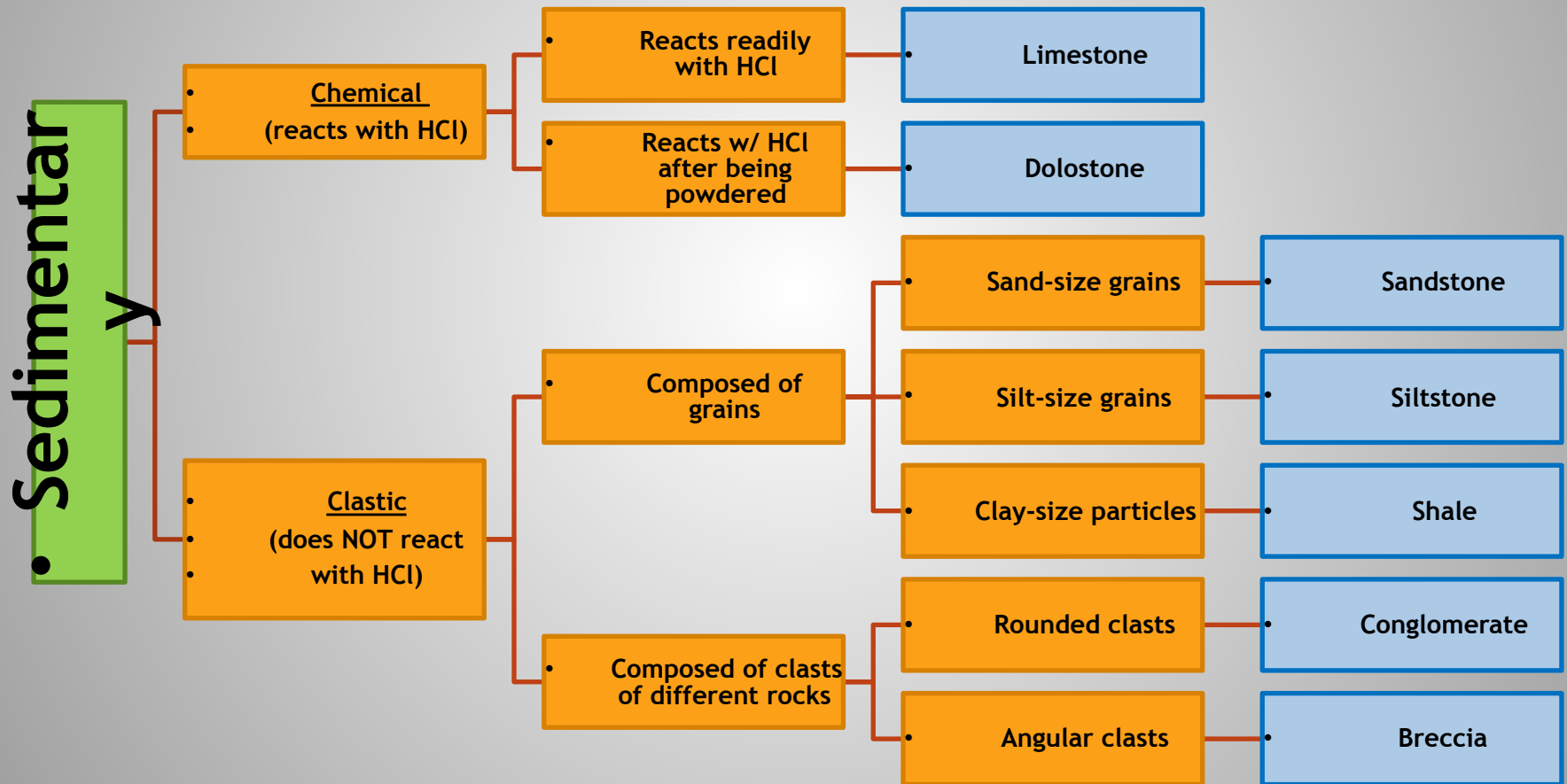
# Flow Chart for Sedimentary Rock Identification

- This is a very comprehensive chart, but is probably a bit too detailed for elementary and middle school Earth science
- The next slide is a much more simplified chart I made that might be a bit more appropriate with rock ID at this grade level



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# Simplified Flow-chart for Identifying Sedimentary Rocks

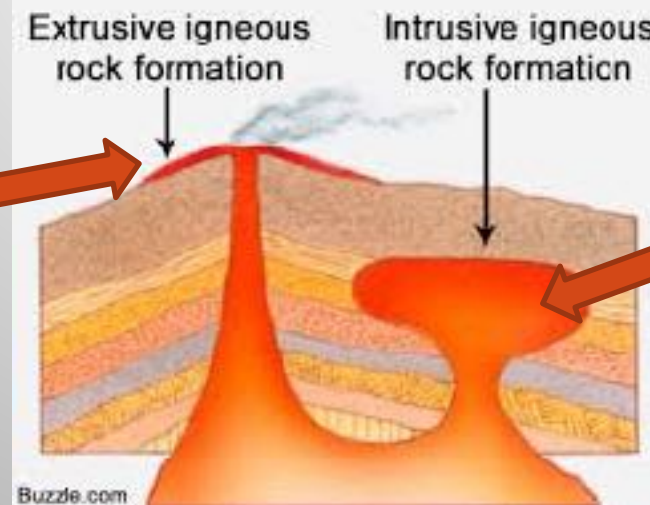


# Igneous Rocks

- Rocks that solidified from a molten (or partially molten) material (magma or lava)
- 2 kinds of igneous rocks
  - Intrusive (form below the earth's surface)
  - Extrusive (form above the earth's surface)



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# Igneous Rock Formation

- How does magma/lava turn into a rock?
- Think about ice forming from water
  - Cools!
  - Solidifies!
    - How does a rock “solidify”???
    - Grows crystals/minerals



# Minerals in Igneous Rocks

- Mineral size is a factor of TIME
  - The more time a rock has to crystallize, the bigger the minerals can grow!
    - Extrusive (Volcanic) = quick cooling, fine-grained
      - *No visible crystals - aphanitic*
    - Intrusive (Plutonic) = slow cooling, coarse-grained
      - *Crystals visible with the naked eye - phaneritic*



**Fine-  
grained**



**Coarse-grained**



# Identifying Igneous Rocks

- Grain size: Fine-, medium-, coarse-grained, or a

combination



fine



medium



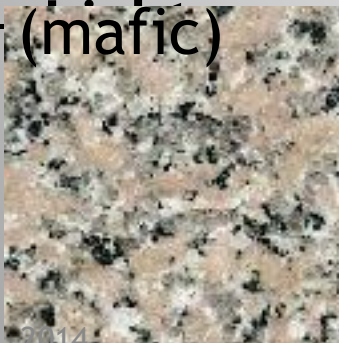
coarse



combination -  
porphyritic)

- Color & Composition: determined by minerals

present  
(felsic)  
- (mafic)



(intermediate)



dark



# Identifying Igneous Rocks (continued)

- Other features
  - Vesicles/Amygdules

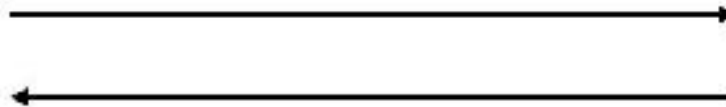


- Glassiness



# Identifying Igneous Rocks (continued)

Increasing Fe and Mg  
 Increasing silica content

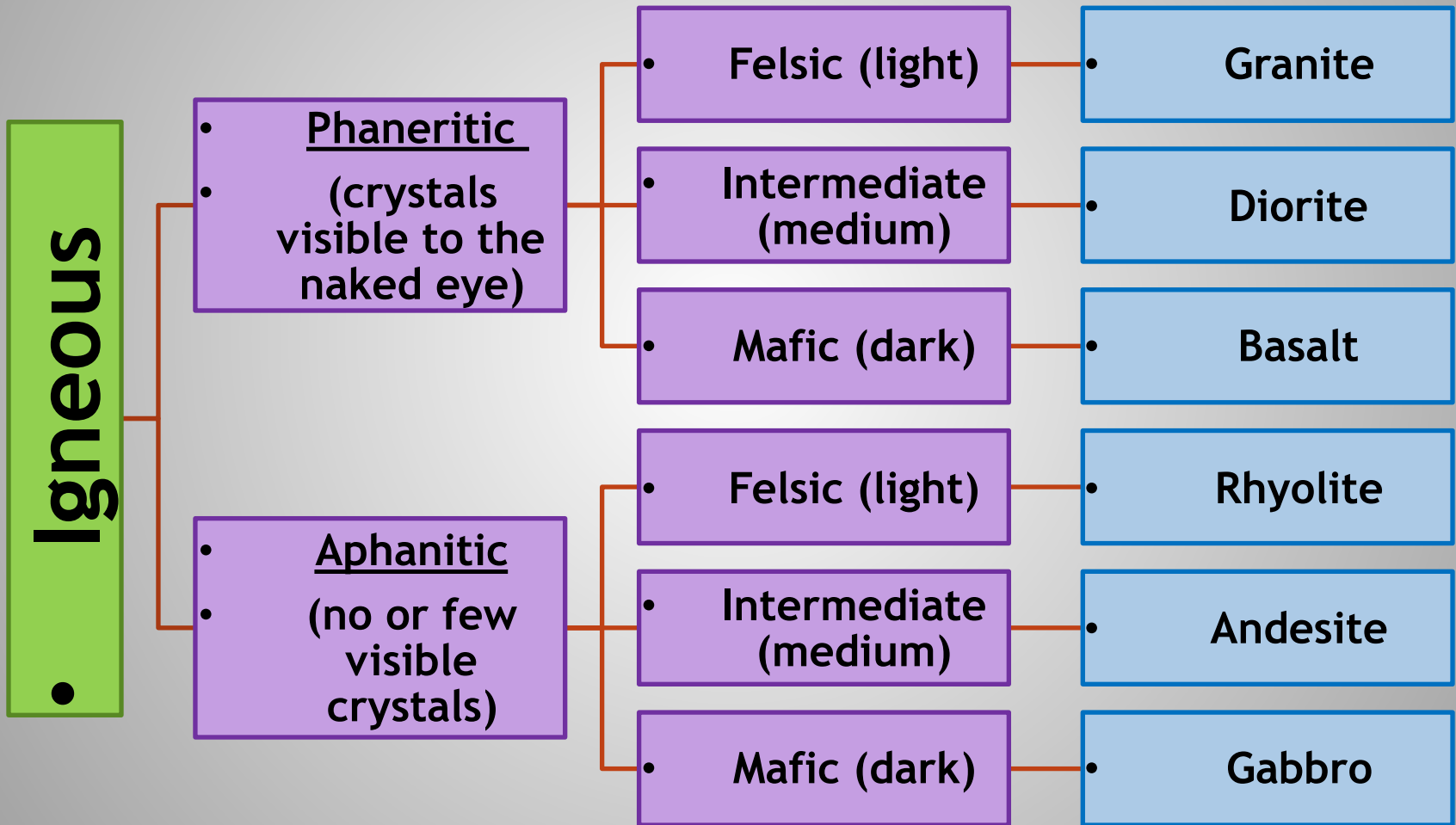


Igneous Rock Classification		Felsic/ Silicic	Intermediate	Mafic
		light colored high in SiO <sub>2</sub>		dark colored low in SiO <sub>2</sub>
Phaneritic (coarse grained)	Plutonic (magma) Intrusive	<b>Granite</b>	<b>Diorite</b>	<b>Gabbro</b>
Aphanitic (fine grained)	Volcanic (lava) Extrusive	<b>Rhyolite</b>	<b>Andesite</b>	<b>Basalt</b>

This table can be VERY useful to help understand the similarities and differences of igneous rocks.

However, a simplified flow chart like I created on the next slide often proves MUCH more helpful in the process of identifying igneous rocks.

# Simplified Flow-chart for Identifying Igneous Rocks



# Metamorphic Rocks

- Metamorphic rocks form when a rock is exposed to elevated temperature and
- **pressure** involves a change in texture or mineralogy in the solid state (*in other words, the rock can't re-melt*)
- Can form from an igneous rock, a sedimentary rock, or another metamorphic rock



# Common metamorphic rocks

- The more heat and pressure you add, the more the rock changes
- The most common structure developed in metamorphic rocks is foliation, when minerals are aligned parallel to one another by deformation



slate



schist



phyllite

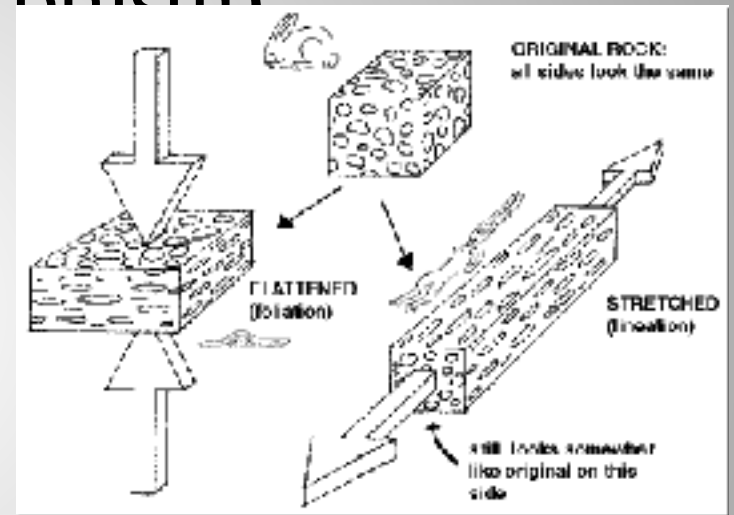
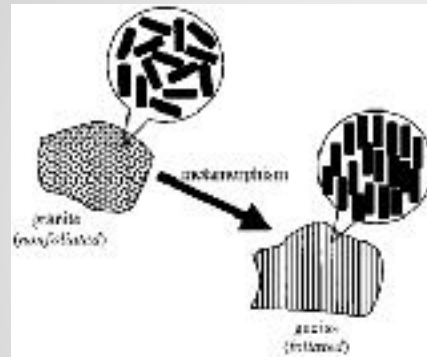


gneiss

# Identifying Metamorphic Rocks

- Grain size (grain size usually increases with increasing metamorphism)

- Foliation



- Other features

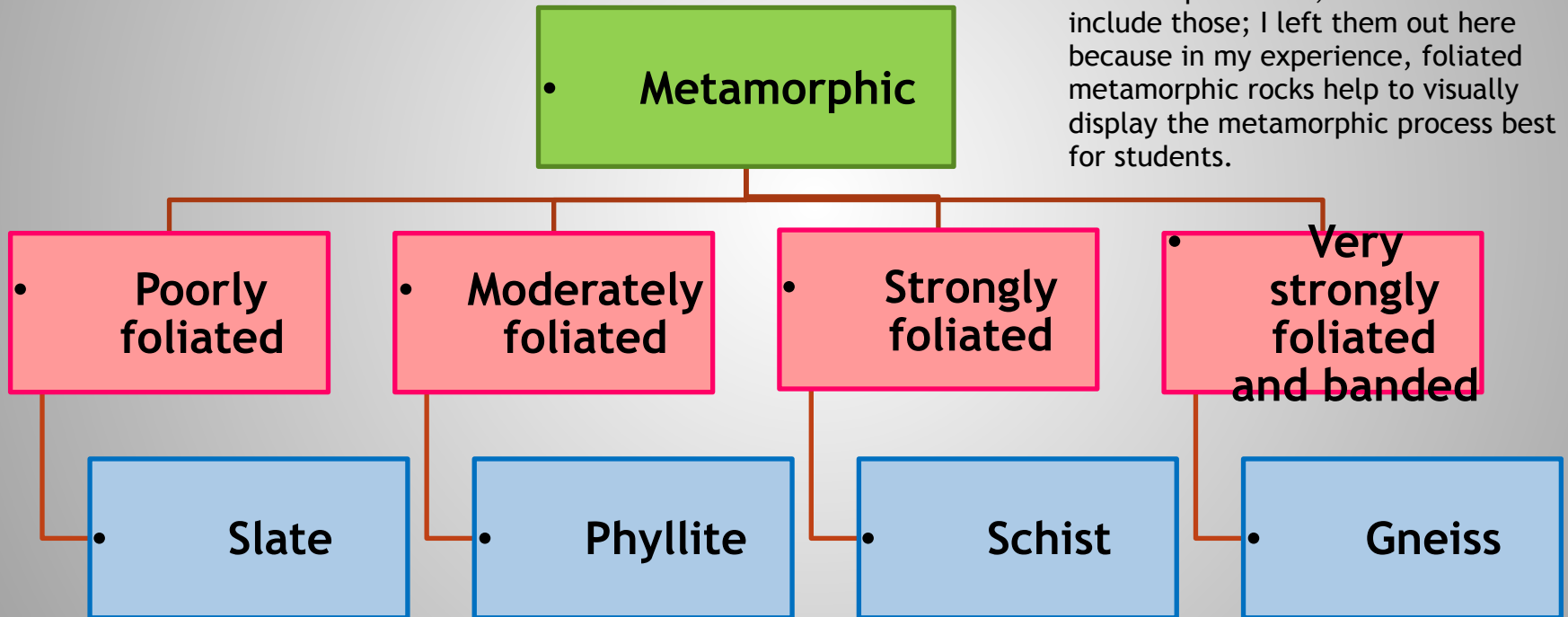
– Folding

“Squished” appearance



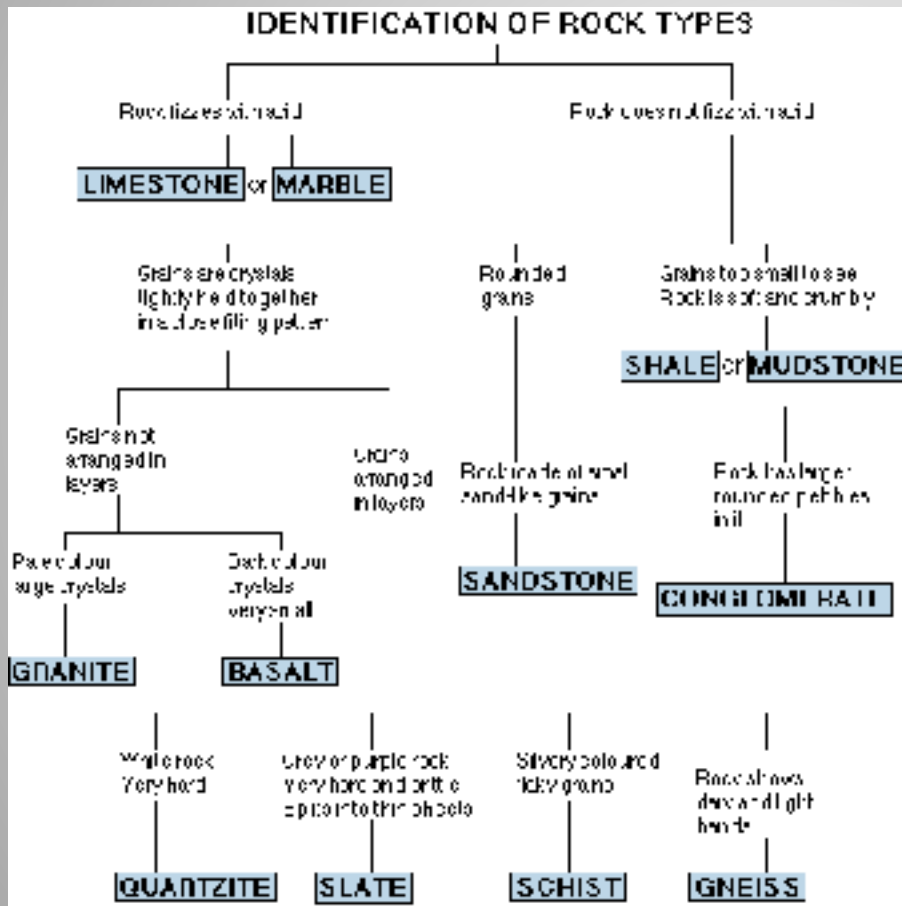
# Simplified Flow-chart for Identifying *Foliated* Metamorphic Rocks

NOTE: Remember, there are non-foliated metamorphic rocks such as quartzite and marble as well. If you have a lot of time to spend on metamorphic rocks, it's wonderful to include those; I left them out here because in my experience, foliated metamorphic rocks help to visually display the metamorphic process best for students.





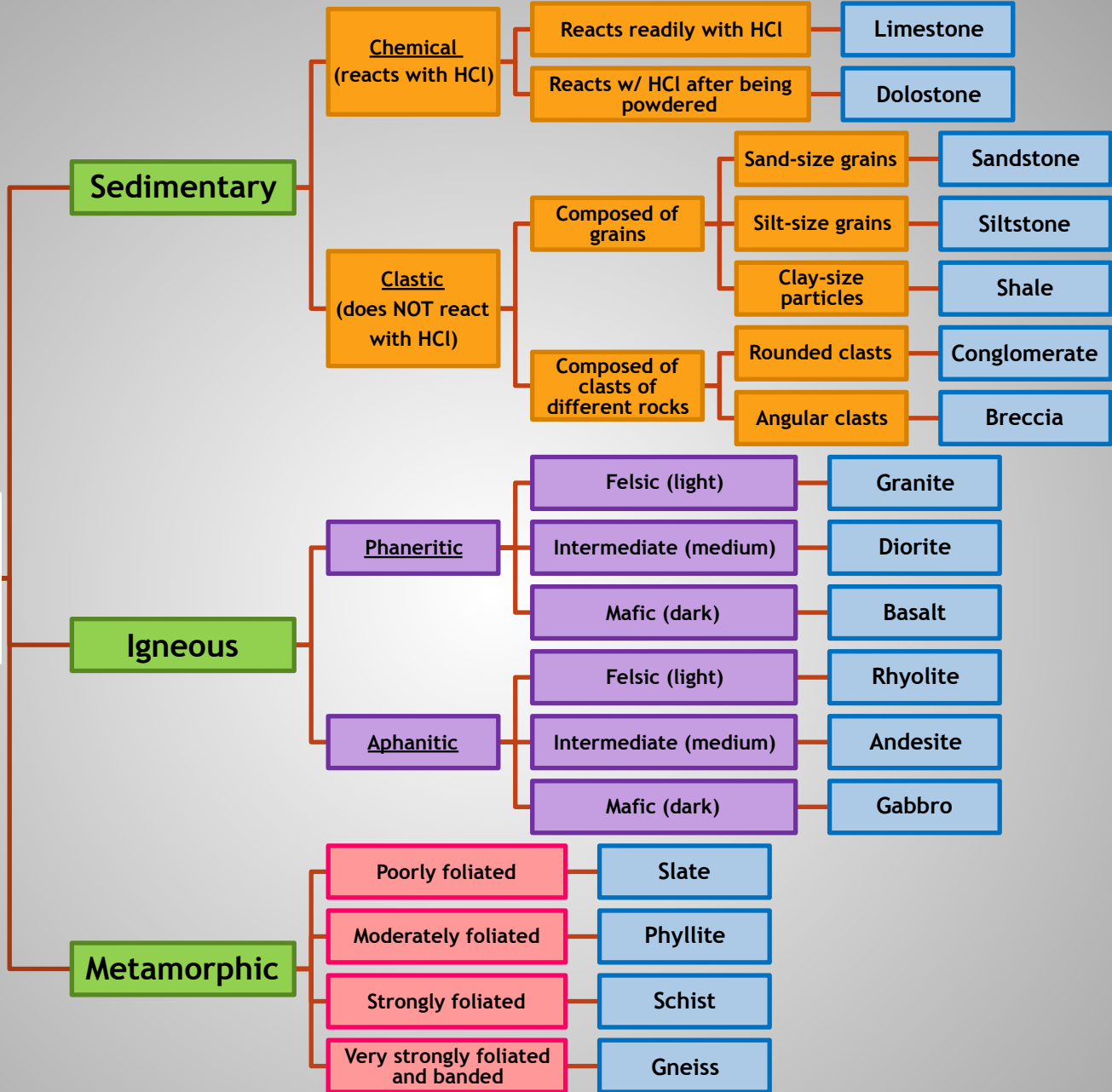
# General Rock Identification



Many rock identification charts like this are available on the Internet, but I found it most helpful to create my own chart that helps students arrive at a rock name by eliminating possibilities in manageable steps (see next slide) ... Hopefully this will help you in your classroom as well!

This rock ID chart is a combination of the three shown previously for the three different rock types.

# What type of rock is it?



Questions?

**LAB TIME!!!**

#	Color	Streak	Hardness	Specific Gravity	Mag	Luster	Cleavage	Mineral Name
0	Grey	Grey	1-1.5	Low	N	Non-metallic	1	graphite
1	Silver	Dk. Grey	2.5-2.75	high		metallic	Cubic (3 @ 90)	Galena
2	Gold	Black/grey	6-6.5	High	weak	Metallic	Cubic (3 @ 90)	Pyrite
3	Dk grey/silver	Dk grey	5.5-6.5	High	y	Metallic	none	Magnetite
4	Red or silver	Red to dk brown	4.5	High	weak	Metallic to non-metallic	none	Hematite
5	Orange to yellow	Orange to yellow	4-5.5	Med-high	N	Non-metallic	none	limonite
6	Clear	white	2-2.5	med	N	Vitreous	2 @ 90	Halite
7	Clear to white	white	1.5-2	High	N	Vitreous, pearly	2 @ 90	Gypsum
8	Clear to yellow	white	3	med	N	Vitreous to pearly	3 not @ 90	Calcite
9	Green	none	6.5-7	high	N	vitreous	none	Olivine
10	Dk green to black	White	5-6	Med	N	Vitreous/shiny to dull	2 @ 90	Pyroxene
11	Dk green to black	white	5-6	Med	N	Vitreous/shiny to dull	2 @60	Hornblende
12	black	White/none	2.5-3	Low	N	shiny	1	Biotite
13	Clear/tan	White/none	2.5-3	Low	N	shiny	1	Muscovite
14	White, tan, grey	None	6	med	N	Non-metallic	2 @ 90	Plagioclase
15	Pink to white	None	6	Med	N	Non-metallic	2 @ 90	Orthoclase